

## ALKALOIDS FROM *HYPERBAENA COLUMBICA*

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(Received 11 February 1983)

**Key Word Index**—*Hyperbaena columbica*; Menispermaceae; alkaloids; 3-demethoxy-2 $\alpha$ ,3 $\alpha$ -methylenedioxyerythroculine; erythroculine; protostephanine.

**Abstract**—Erythroculine, protostephanine and a new alkaloid have been isolated from *Hyperbaena columbica*, and the latter has been identified as 3-demethoxy-2 $\alpha$ ,3 $\alpha$ -methylenedioxyerythroculine.

### INTRODUCTION

In *Hyperbaena columbica* (Eichl.) Miers three alkaloids have been detected, but not identified [1]. According to the optical rotation [2], IR [2], UV [2] and  $^1\text{H}$  NMR [2] as well as the mass spectral fragmentation (cf. [3]), one alkaloid is identical to erythroculine (1). According to melting point [4], UV [5],  $^1\text{H}$  NMR and the high resolution mass spectrum the second alkaloid is protostephanine (3). Comparison of the IR, UV and ORD spectra (positive Cotton effect at 300 nm) of the third alkaloid with those of erythroculine (1) indicated a similar structure including absolute configuration.

### RESULTS AND DISCUSSION

$^1\text{H}$  NMR and mass spectrometry (Fig. 1) showed that the new alkaloid has the structure 2. Spin-spin decoupling established the partial structure  $\text{R}^1\text{R}^2\text{R}^3\text{C}-\text{CH}_2-\text{CH}(\text{OR}^4)-\text{CH}(\text{OR}^5)-\text{CH}=\text{CR}^6\text{R}^7$ . The coupling constants  $J_{3,4\alpha} = 11.5$  Hz and  $J_{3,4\beta} = 6.0$  Hz proved the axial conformation of H-3 and hence its 3 $\beta$ -configuration. NOE enhancements of 6 and 11% were observed between H-2 (H-14) and H-3 (H-14), respectively. Such small positive values occur for the acute angle case of a three-spin system [6], pointing to the 2 $\alpha$ ,3 $\alpha$ -configuration of the methylenedioxy group. In the case of a 2 $\beta$ ,3 $\alpha$ -configuration, a negative NOE enhancement between H-2 (H-14) would be expected.

*Note added in proof:* Considering additionally H-1, H-4 $\alpha$  and H-4 $\beta$  a negative NOE enhancement between H-2 (H-14) for the 2 $\beta$ ,3 $\alpha$ -configuration was also calculated.

### EXPERIMENTAL

**Plant material.** *H. columbica* (Eichl.) Miers was collected in May in Sierra del Rosario, Pinar del Río, Cuba, and identified by Lic. Pedro Herrera, Havana. A voucher specimen is retained in the Herbarium of the Institute of Botany, Academy of Sciences of Cuba, Havana.

**NOE studies.** These were performed on a degassed sample of less than 5% concn (w/v) in  $\text{Me}_2\text{CO}-d_6$ . Each resonance was integrated 7 times in a homo-inverse-gated decoupled and a decoupled spectrum and the NOEs were calculated from the average areas. Accuracies of NOE values are  $\pm 1.5\%$ .

**Extraction.** Dried (40°) and ground leaves were extracted with EtOH at room temp. Evapn of the EtOH *in vacuo* gave a residue which was partitioned between 0.5 M HCl and  $\text{C}_6\text{H}_6-\text{Et}_2\text{O}$  (1:1). After addition of  $\text{KHCO}_3$  to the aq. layer, the latter was extracted with  $\text{CHCl}_3-\text{EtOH}$  (2:1). Evapn of the solvent gave raw material which was chromatographed over silica gel using  $\text{CHCl}_3-\text{MeOH}$  mixtures.

**3-Demethoxy-2 $\alpha$ ,3 $\alpha$ -methylenedioxyerythroculine (2).** Elution with  $\text{CHCl}_3-\text{MeOH}$  (99:1) and crystallization from MeOH afforded 2; yield 0.003%; mp 94–97°,  $[\alpha]_D^{25} + 229^\circ$  ( $\text{CHCl}_3$ ; c 0.56). IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 1732 ( $\text{CO}_2\text{R}$ ), 1612, 1500 (aromatic ring). UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 305 (3.54), 238 (4.07, sh), 215 (4.55). ORD (EtOH):  $[\phi]_{315} + 2400^\circ$  (peak),  $[\phi]_{294} + 2100^\circ$  (trough).  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ , TMS):  $\delta$  1.55 (t,  $J = 11.5$  Hz, 1H, H-4 $\alpha$ ), 2.27 (dd,  $J = 11.5$  and 6.0 Hz, 1H, H-4 $\beta$ ), 3.83 (s, 3H, OMe), 3.85 (s, 3H, OMe), 4.38 (m, 1H, H-3), 4.60 (m, 1H, H-2), 4.95 and 5.11 (each s, each 1H,  $\text{OCH}_2\text{O}$ ), 5.90 (m, 1H, H-1), 6.68 (s, 1H, H-17), 7.26 (s, 1H, H-14). MS 70 eV,  $m/z$  (rel. int.): 357.1569 (calc. for  $\text{C}_{20}\text{H}_{23}\text{NO}_5$ : 357.1576; 16), 342.1340 (calc. for  $\text{C}_{19}\text{H}_{20}\text{NO}_5$ : 342.1341; 0.4), 326.1384 (calc. for  $\text{C}_{19}\text{H}_{20}\text{NO}_4$ : 326.1392; 7), 300.1230 (calc. for  $\text{C}_{17}\text{H}_{18}\text{NO}_4$ : 300.1236; 100), 285.1360 (calc. for  $\text{C}_{17}\text{H}_{19}\text{NO}_3$ : 285.1365; 83).

**Erythroculine (1).** Elution with  $\text{CHCl}_3-\text{MeOH}$  (49:1) and crystallization from  $\text{Et}_2\text{O}$ -pentane afforded 1; yield 0.03%; mp 79–82°,  $[\alpha]_D^{25} + 216^\circ$  ( $\text{CHCl}_3$ ; c 1.05), lit. [2]:  $[\alpha]_D + 194^\circ$  ( $\text{CHCl}_3$ ). IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 1716 ( $\text{CO}_2\text{R}$ ), 1611, 1500 (aromatic ring). UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 303 (3.49), 238 (3.93, sh), 214 (4.38). ORD (EtOH):  $[\phi]_{315} + 1250^\circ$  (peak);  $[\phi]_{296} + 1150^\circ$  (trough).  $^1\text{H}$  NMR (100 MHz,  $\text{CDCl}_3$ , TMS):  $\delta$  1.57 (t,  $J = 12$  Hz, 1H, H-4 $\alpha$ ), 2.28 (dd,  $J = 12$  and 4 Hz, H-4 $\beta$ ), 3.29 (s, 3H, 3-OMe), 3.89 (s, 6H, 16-OMe,  $\text{CO}_2\text{Me}$ ), 5.62 (m, 1H, H-1), 6.71 (s, 1H, H-17), 7.52 (s, 1H, H-14). MS 70 eV,  $m/z$  (rel. int.): 343.1780 (calc. for  $\text{C}_{20}\text{H}_{25}\text{NO}_4$ : 343.1783; 7), 328.1556 (calc. for  $\text{C}_{19}\text{H}_{22}\text{NO}_4$ : 328.1549; 0.5), 312.1602 (calc. for  $\text{C}_{19}\text{H}_{22}\text{NO}_3$ : 312.1600; 13), 285.1366 (calc. for  $\text{C}_{17}\text{H}_{19}\text{NO}_3$ : 285.1365; 100).

**Protostephanine (3).** Further elution with  $\text{CHCl}_3-\text{MeOH}$  (49:1) and crystallization from MeOH afforded 3; yield 0.01%; mp 52–54°, lit. [4]: mp 75°. IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1603, 1513 (aromatic ring). UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (log  $\epsilon$ ): 283 (3.92).  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ , TMS):  $\delta$  2.42 (s, 3H, NMe), 3.87 (s, 3H, OMe), 3.89 (s, 6H, OMe), 3.99 (s, 3H, OMe), 6.41 (d,  $J = 3$  Hz, 1H), 6.54 (d,  $J = 3$  Hz, 1H), 6.74 (s, 1H), 6.82 (s, 1H) (aromatic protons). MS 70 eV,  $m/z$  (rel. int.): 357.1932 (calc. for  $\text{C}_{21}\text{H}_{27}\text{NO}_4$ : 357.1940; 100), 342.1702 (calc. for  $\text{C}_{20}\text{H}_{24}\text{NO}_4$ : 342.1705; 18), 326.1756 (calc. for  $\text{C}_{20}\text{H}_{24}\text{NO}_3$ : 326.1756; 20).

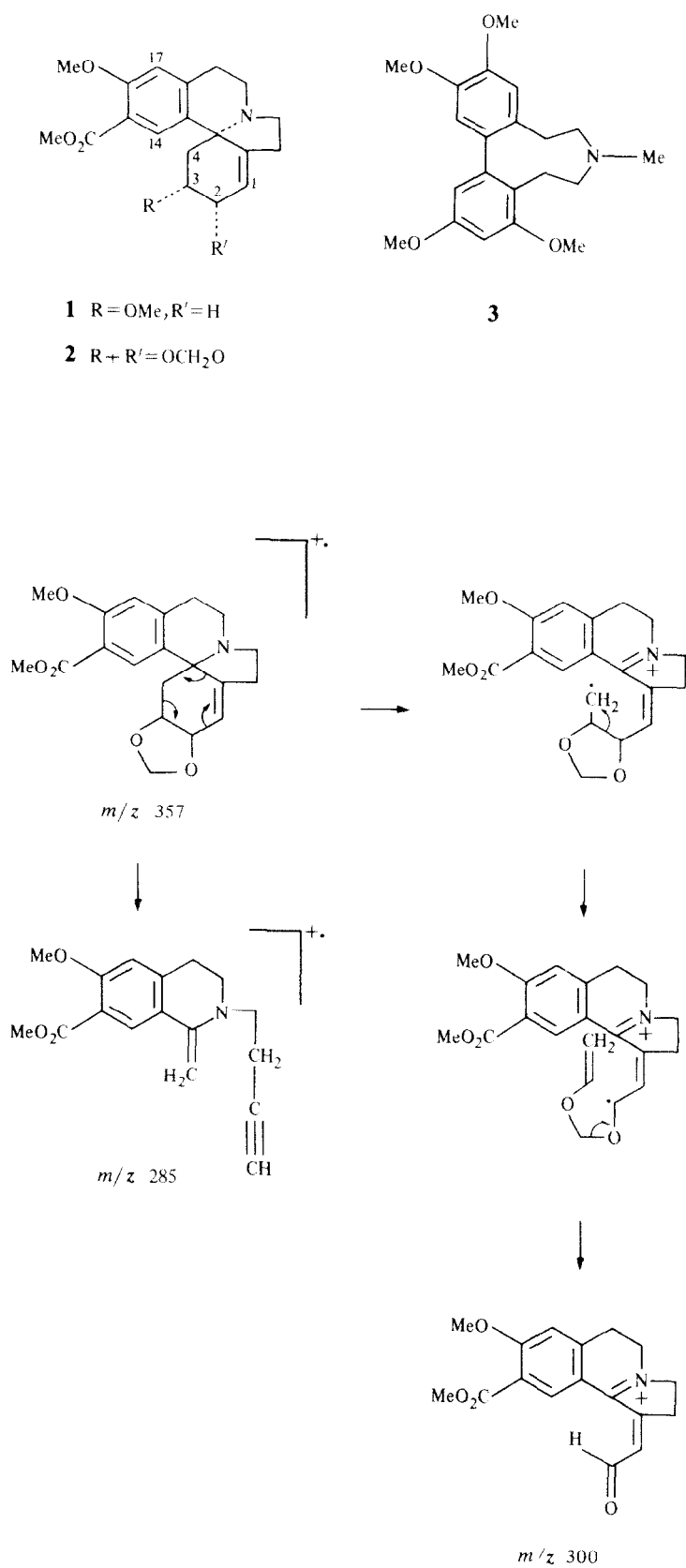


Fig. 1. MS fragmentation of 3-demethoxy-2 $\alpha$ ,3 $\alpha$ -methylenedioxyerythroculine (cf. [3]).

*Acknowledgement*—We thank Dr. A. Zschunke, Halle, for helpful discussions.

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